

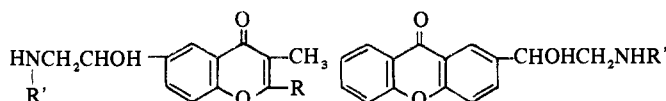
Notes

 β -Adrenergic Blocking Agents of the Chromone and Xanthone Groups

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The search for new β -adrenergic blocking drugs with enhanced therapeutic properties is still very active¹ and there has recently been an increasing interest in heterocyclic (e.g., benzodioxane,^{2,3} indole,^{4,5} benzofuran,⁶ dibenzofuran,⁷ quinoline, etc.⁸) derivatives. We now report the preparation of some new β -adrenergic blocking derivatives of the chromone and flavone group, with which we wish to illustrate further the versatility of the benzo- γ -pyrone molecule as a carrier moiety in medicinal research. We have also prepared 2 analogous xanthone derivatives (**19**, **20**) on the basis of the results obtained among the CNS stimulants of the same group.^{9,10} The basic chain was located, according to the synthetic possibilities, in position 6 for **4**, **6**, **10**, and **12**



- 4**, R = CH₃; R' = *i*-C₃H₇
6, R = C₆H₅; R' = *i*-C₃H₇
10, R = CH₃; R' = *tert*-C₄H₉
12, R = C₆H₅; R' = *tert*-C₄H₉
- 19**, R' = *i*-C₃H₇
20, R' = *tert*-C₄H₉

and in the 2 position for **19** and **20**. The synthesis of these compds was carried out from the corresponding Ac derivatives, prepared by standard methods, by bromination, amination, and reduction of the final amino ketone intermediates. As an alternative, the reductive amination of the glyoxalyl derivatives (obtained by SeO₂ oxidation of the same starting materials) according to Fodor and Kovacs¹¹ gave unsatisfactory results. In Table I all the chromone and flavone derivatives prepared are shown, while the xanthone

analogues are described in detail in the Experimental Section for illustrative purposes.

Compds **4**, **6**, **10**, **12**, **19**, and **20** have been evaluated for their β -adrenergic blocking activity in a number of tests at the Institute of Pharmacology of the University of Padua. The *N*-isopropyl derivatives were significantly more active than the *tert*-Bu analogs; the activity of the former group decreases in the order **4** > **19** > **10**. In this group, 6-[1-hydroxy-2-isopropylaminoethyl]-2,3-dimethylchromone (**4**), was the most active in all tests employed.

(a) **Lipomobilizing Activity.**¹² The norepinephrine-induced lipid mobilization in rats is competitively blocked by equimolecular doses of **4**, down to $2 \times 10^{-5} M$.

(b) **Isolated Short-Circuited Frog Skin.**^{13,14} The increase of the short-circuited current in isolated frog skin is antagonized (60%) by **4** at $10^{-6} M$.

(c) **Contraction Rates of Isolated Right Atrial Strips.**¹⁵ The chronotropic effect of 0.001 $\mu g/ml$ of isoproterenol is reduced to about half and completely abolished with doses of **5** and, respectively, 10 $\mu g/ml$ of **4**.

(d) **Lengthening of the Refractory Period of the Isolated Guinea Pig Auricles.**¹⁶ An average 20% (EC₂₀) reduction in the maximal driven rate (MDR) at which the isolated guinea pig auricles respond to electrical stimulation is produced by 10 $\mu g/ml$ of **4**. On the basis of the EC₂₀ values, propranolol is approximately 10 times more active.

(e) **Isolated Guinea Pig Tracheal Chain.**¹⁷ The isoproterenol reduction of the contraction of the isolated guinea pig tracheal chain caused by carbachol (0.5 $\mu g/ml$) is antagonized by very small doses of **4**. Figure 1 reports cumulative log concn response curves for the agonist (isoproterenol) in the presence of various concns of the antagonist (**4**).

(f) **Blood Pressure in Anesthetized Dogs.**¹⁵ Iv administration of 1 and 5 mg/kg of **4** causes a 100% and 87% reduction of the hypotensive response to 0.2 and, respectively, 1 $\mu g/kg$ of iv isoproterenol. The same doses of **4** cause a slight and transient hypotensive response of the order of -5 and -10 mm.

Compd **4** is therefore a selective β -adrenergic blocking

Table I. Chromone and Flavone Derivatives

Compd ^d	R	R'	X	Mp, °C	Formula	Analyses
1	CH ₃	H	CO	136-138 ^a	C ₁₃ H ₁₂ O ₃	C, H
2	CH ₃	Br	CO	143-146 ^b	C ₁₃ H ₁₁ BrO ₃	C, H, Br
3	CH ₃	NH- <i>i</i> -C ₃ H ₇ · HCl	CO	220-223 ^c	C ₁₆ H ₂₀ ClNO ₃	C, H, Cl, N
4	CH ₃	NH- <i>i</i> -C ₃ H ₇ · HCl	CHOH	233-235 ^c	C ₁₆ H ₂₂ ClNO ₃	C, H, Cl, N
5	CH ₃	NH- <i>tert</i> -C ₄ H ₉ · HCl	CO	200-202 ^c	C ₁₇ H ₂₂ ClNO ₃	C, H, Cl, N
6	CH ₃	NH- <i>tert</i> -C ₄ H ₉ · HCl	CHOH	244-247 ^c	C ₁₇ H ₂₄ ClNO ₃	C, H, Cl, N
7	C ₆ H ₅	H	CO	154-156 ^a	C ₁₈ H ₁₄ O ₃	C, H
8	C ₆ H ₅	Br	CO	216-217 ^b	C ₁₈ H ₁₃ BrO ₃	C, H, Br
9	C ₆ H ₅	NH- <i>i</i> -C ₃ H ₇ · HCl	CO	137-139 ^c	C ₂₁ H ₂₂ ClNO ₃	C, H, Cl, N
10	C ₆ H ₅	NH- <i>i</i> -C ₃ H ₇ · HCl	CHOH	202-204 ^c	C ₂₁ H ₂₄ ClNO ₃	C, H, Cl, N
11	C ₆ H ₅	NH- <i>tert</i> -C ₄ H ₉ · HCl	CO	257-259 ^c	C ₂₂ H ₂₄ ClNO ₃	C, H, Cl, N
12	C ₆ H ₅	NH- <i>tert</i> -C ₄ H ₉ · HCl	CHOH	269-270 ^c	C ₂₂ H ₂₆ ClNO ₃	C, H, Cl, N

Crystn solvent: ^aligroin, ^bEtOAc, ^cMeOH-Et₂O. ^dThe bases corresponding to compds **4**, **6**, **10**, and **12** melted respectively (ligroin) at 105-107°, 157-159°, 121-122°, 146-148°.

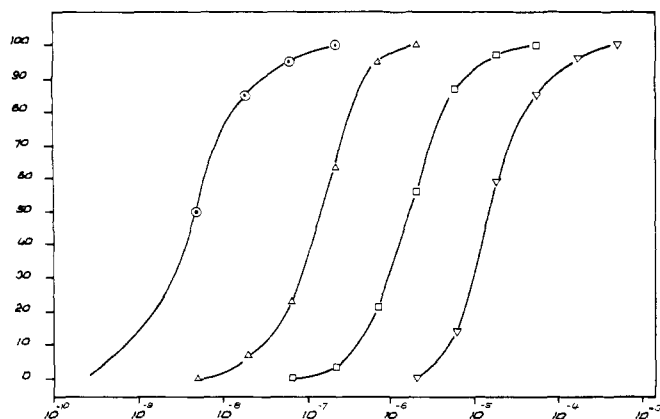


Figure 1. Cumulative log concentration-response curves for isoproterenol in the presence of various concns (\circ , 0 M; \triangle , 3.2×10^{-7} M; \square , 3.2×10^{-6} M; ∇ , 3.2×10^{-5} M) of 6-[1-hydroxy-2-isopropylaminoethyl]-2,3-dimethylchromone. Abscissa: molar concns of isoproterenol. Ordinate: per cent of maximal release.

agent of the propranolol type,¹⁸ with membrane activity (test d) and devoid of intrinsic sympathomimetic activity (test f). In comparison with propranolol, 4 has a potency ratio of 0.1 (test d), but its LD₅₀ [223 mg/kg (205.5–242.0) iv in albino mice] is 2.5 times lower.

The data reported seem to confirm the pharmacological potentialities of the benzo- γ -pyrone molecule also in this field of medicinal chemistry.

Experimental Section

All melting points were detd in open glass capillaries, using a Büchi apparatus, and are uncor. Where analyses are indicated only by symbols of the elements, analytical results obtained for those elements were within $\pm 0.4\%$ of the theoretical values.

The general methods of synthesis described are illustrative of those of analogous compounds.

2-Propionyl-4-acetylphenol (13). To a soln of 25 ml of *o*-hydroxypropionophenone and 25 ml of AcCl in 100 ml of CS₂, 95 g of AlCl₃ was added in 1 hr, and the mixt was kept at 50–60° for 1 hr. After removal of the solvent the residue was treated with ice and HCl and extd with CHCl₃. The CHCl₃ layer was dried and filtered, and the solvent was evapd. The residue was distd at 175–180° (4–5 mm); the collected oil solidified on cooling. On crystg from ligroin, 23 g of white product (mp 64–65°) was obtd. *Anal.* (C₁₁H₁₂O₃) C, H.

6-Acetyl-2,3-dimethylchromone (1). A mixt of 10 g of 2-propionyl-4-acetylphenol (13), 6 g of anhyd NaOAc, and 10 ml of Ac₂O was heated in an oil bath at 170–180° for 7 hr. The fused mixt was taken up in H₂O and the sepd solid was collected, washed (H₂O), and dried. On crystg from ligroin, 8 g of yellow product, mp 136–138°, was obtd. *Anal.* (C₁₃H₁₄O₃) C, H.

6-Acetyl-3-methylflavone (7). A mixt of 15 g of 2-propionyl-4-acetylphenol (13), 30 g of BzCl, and 45 g of PhCOONa was heated in an oil bath at 180–190° for 7–8 hr. The reaction mixt was taken up in H₂O, washed (NaOH–H₂O), and extd with CHCl₃. Removal of the solvent left a residue which on crystg from ligroin, gave 15 g of yellow solid, mp 154–156°. *Anal.* (C₁₈H₁₄O₃) C, H.

2-Carboxy-4'-acetyldiphenyl Ether (14). A mixt of 7.8 g of *o*-chlorobenzoic acid, 6 g of *p*-hydroxyacetophenone, 4 g of NaOH, and 0.5 g of Cu powder was heated in an oil bath at 200° for 0.5 hr. The mixt was poured into ice water and acidified with dil HCl, and the sepd product was isolated by filtration, dissolved (NaHCO₃), and repptd with dil HCl. The crude product, on crystg from EtOH–H₂O, gave 6 g of white cryst solid, mp 185–186°. *Anal.* (C₁₅H₁₂O₄) C, H.

2-Acetyl-xanthone (15). To a soln of polyphosphoric acid (from 100 g of P₂O₅ and 100 g of 85% H₃PO₄), 9.3 g of 14 was added in small portions. The reaction mixt was kept on a steam

bath for 1.5 hr and then poured into ice H₂O. The sepd solid was collected, washed (NaHCO₃–H₂O), and dried. On crystg from EtOH, 5.5 g of white solid, mp 199–200°, was obtd. *Anal.* (C₁₅H₁₀O₃) C, H.

2-Bromoacetyl-xanthone (16). To a soln of 4.8 g of 15 in 300 ml of CHCl₃, a soln of 3.2 g of Br₂ in 75 ml of CHCl₃ was added, with stirring, in 2 hr. The reaction mixt was transferred to a separatory funnel, washed (dil NaOH–H₂O), and dried, and the solvent was evapd. The residue, after 2 crystns from EtOAc, gave 3 g of white solid, mp 194–196°. *Anal.* (C₁₅H₉BrO₃) C, H, Br.

2-Isopropylaminoacetyl-xanthone Hydrochloride (17). To a soln of 17 g of 16 in 1.5 l. of PhH, a slight excess of *i*-PrNH₂ was added, and the mixt was kept at room temp with stirring for 4 hr. The soln was washed (H₂O) and dried. The PhH layer, treated with HCl gas, gave the amino ketone as the HCl salt. On crystallg the crude product from MeOH–Et₂O, 8 g of white solid, mp 210–211°, was obtd. *Anal.* (C₁₈H₁₈ClNO₃) C, H, Cl, N.

2-*tert*-Butylaminoacetyl-xanthone Hydrochloride (18). In a similar manner, starting from 7.3 g of 2-bromoacetyl-xanthone (16), 3 g of the corresponding amino ketone hydrochloride (18), mp 189–193° dec (MeOH–Et₂O), was obtained. *Anal.* (C₁₉H₂₀ClNO₃) C, H, Cl, N.

2-[1-Hydroxy-2-isopropylaminoethyl]-xanthone Hydrochloride (19). A soln of 4.1 g of 17 in 40 ml of MeOH was hydrogenated over 10% Pd/C until H₂ uptake ceased. The soln was filtered from the catalyst and evapd to dryness. The residue, on crystg from MeOH–Et₂O gave 2.5 g of white product, mp 224–226° [*Anal.* (C₁₈H₂₀ClNO₃) C, H, Cl, N]; base, white cryst solid, mp 108–110° (ligroin) [*Anal.* (C₁₈H₁₉NO₃) C, H, N].

2-[1-Hydroxy-2-*tert*-butylaminoethyl]-xanthone Hydrochloride (20). With the same procedure 3 g of *tert*-butylaminoacetyl-xanthone hydrochloride (18) gave 1.7 g of the corresponding amino alcohol hydrochloride as a white solid, mp 230–233° (MeOH–Et₂O); [*Anal.* (C₂₀H₂₂ClNO₃) C, H, Cl, N]; base, white cryst solid, mp 112–114° (ligroin) [*Anal.* (C₁₉H₂₁NO₃) C, H, N].

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References

- (1) M. S. K. Ghouri and T. J. Haley, *J. Pharm. Sci.*, **58**, 511 (1969).
- (2) G. Marchetti, L. Merlo, and V. Nosedà, *Arzneim.-Forsch.*, **18**, 43 (1968).
- (3) R. Howe, B. S. Rao, and M. S. Chodnekar, *J. Med. Chem.*, **13**, 169 (1970).
- (4) J. H. Biel and B. K. B. Lum, *Fortschr. Arzneimittelforsch.*, **10**, 46 (1966).
- (5) R. C. Hill and P. Turner, *Brit. J. Pharmacol.*, **36**, 368 (1969).
- (6) R. C. Hill and P. Turner, *Brit. J. Pharmacol. Chemother.*, **32**, 663 (1968).
- (7) R. Wandestrück, C. Goldenberg, F. Binon, and R. Charlier, *Chim. Ther.*, **1970**, 285.
- (8) W. Hepworth, A. Mitchell, N. S. Chodnekar, and R. Howe, French Patent 3697 (Dec 27, 1965); *Chem. Abstr.* 82119 (1967).
- (9) P. Da Re, V. Mancini, E. Toth, and L. Cima, *Arzneim.-Forsch.*, **18**, 718 (1968).
- (10) P. Da Re, L. Sagradora, V. Mancini, P. Valenti, and L. Cima, *J. Med. Chem.*, **13**, 527 (1970).
- (11) G. Fodor and O. Kovacs, *J. Amer. Chem. Soc.*, **71**, 1045 (1949).
- (12) G. Fassina, *Arch. Int. Pharmacodyn.*, **166**, 281 (1967).
- (13) H. H. Ussig and K. Zerahn, *Acta Physiol. Scand.*, **23**, 110 (1951).
- (14) G. Fassina, F. Carpenedo, and G. Fiandini, *J. Pharm. Pharmacol.*, **20**, 240 (1968).
- (15) J. W. Blank, W. A. M. Duncan, and R. G. Shanks, *Brit. J. Pharmacol.*, **25**, 577 (1965).
- (16) F. P. Luduena, J. H. Howard, and J. K. Borland, *Arch. Int. Pharmacodyn.*, **107**, 335 (1956).
- (17) P. N. Patil, *J. Pharmacol. Exp. Ther.*, **160**, 308 (1968).
- (18) J. D. Fitzgerald, *Clin. Pharmacol. Ther.*, **10**, 292 (1969).